

# **DUNLOP Concrete Resurfacer Tint – Charcoal**

# Ardex (Ardex Australia)

Chemwatch: **52-0572** Version No: **5.1.1.1** 

Safety Data Sheet according to WHS and ADG requirements

#### Chemwatch Hazard Alert Code: 3

Issue Date: 10/08/2015 Print Date: 12/08/2015 Initial Date: Not Available S.GHS.AUS.EN

# SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

#### **Product Identifier**

Product name	DUNLOP Concrete Resurfacer Tint – Charcoal	
Synonyms	Not Available	
Proper shipping name	PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)	
Other means of identification	Not Available	

#### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Use according to manufacturer's directions.
	Colouring of solvent based industrial paints and coatings.

# Details of the manufacturer/importer

Registered company name	Ardex (Ardex Australia)	Ardex (Ardex NZ)
Address	20 Powers Road Seven Hills 2147 NSW Australia 32 Lane Street Woolston Christchurch New Zealand	
Telephone	1800 224 070	+64 3384 3029
Fax	Fax +61 2 9838 7817 +64 3384 9779	
Website	Not Available	Not Available
Email	Not Available	Not Available

# Emergency telephone number

Association / Organisation	Not Available Not Available	
Emergency telephone numbers	1800 224 070 (Mon-Fri, 9am-5pm)	+64 3373 6900
Other emergency telephone numbers	Not Available	Not Available

# **SECTION 2 HAZARDS IDENTIFICATION**

#### Classification of the substance or mixture

#### HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the Model WHS Regulations and the ADG Code.

#### CHEMWATCH HAZARD RATINGS

	Min	Max	
Flammability	2		
Toxicity	2		0 = Minimum
Body Contact	3		1 = Low 2 = Moderate
Reactivity	1		3 = High
Chronic	2		4 = Extreme

Poisons Schedule	S6	
GHS Classification [1]	Flammable Liquid Category 3, Acute Toxicity (Oral) Category 4, Acute Toxicity (Dermal) Category 4, Acute Toxicity (Inhalation) Category 4, Skin Corrosion/Irritation Category 2, Serious Eye Damage Category 1, Carcinogen Category 2, STOT - RE Category 2, Aspiration Hazard Category 1, Acute Aquatic Hazard Category 2, Chronic Aquatic Hazard Category 2	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HSIS; 3. Classification drawn from EC Directive 1272/2008 - Annex VI	

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**GHS** label elements











SIGNAL WORD DANGER

#### Hazard statement(s)

H226	Flammable liquid and vapour	
H302	Harmful if swallowed	
H312	Harmful in contact with skin	
H332	Harmful if inhaled	
H315	Causes skin irritation	
H318	Causes serious eye damage	
H351	Suspected of causing cancer	
H373	May cause damage to organs through prolonged or repeated exposure	
H304	May be fatal if swallowed and enters airways	
H401	Toxic to aquatic life	
H411	Toxic to aquatic life with long lasting effects	
AUH066	Repeated exposure may cause skin dryness and cracking	

#### Precautionary statement(s) Prevention

P201	Obtain special instructions before use.	
P210	10 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.	
P233	Keep container tightly closed.	
P260	P260 Do not breathe dust/fume/gas/mist/vapours/spray.	

#### Precautionary statement(s) Response

P301+P310	IF SWALLOWED: Immediately call a POISON CENTER/doctor/physician/first aider	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P308+P313	IF exposed or concerned: Get medical advice/attention.	
P331	Do NOT induce vomiting.	

#### Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.	
P405	Store locked up.	

### Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration

# SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### Substances

See section below for composition of Mixtures

#### **Mixtures**

CAS No	%[weight]	Name
20344-49-4	0-35	<u>ferric hydroxide</u>
1330-20-7	12.5-17.5	<u>xylene</u>
Not Available	14-18	acrylic polymer
100-41-4	<5	<u>ethylbenzene</u>
108-65-6	5-7.5	propylene glycol monomethyl ether acetate, alpha-isomer
70657-70-4	<0.1	propylene glycol monomethyl ether acetate, beta-isomer
Not Available	30-60	Other organic/mineral pigments

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

# **SECTION 4 FIRST AID MEASURES**

# Description of first aid measures

Eye Contact

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	If this product comes in contact with the eyes:  Immediately hold eyelids apart and flush the eye continuously with running water.  Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.  Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.  Transport to hospital or doctor without delay.  Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs:  Immediately remove all contaminated clothing, including footwear.  Flush skin and hair with running water (and soap if available).  Seek medical attention in event of irritation.
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor.</li> </ul>
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> <li>Avoid giving milk or oils.</li> <li>Avoid giving alcohol.</li> <li>If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus.</li> </ul>

#### Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

Treat symptomatically.

For acute or short term repeated exposures to iron and its derivatives:

- Always treat symptoms rather than history.
- In general, however, toxic doses exceed 20 mg/kg of ingested material (as elemental iron) with lethal doses exceeding 180 mg/kg.
- Control of iron stores depend on variation in absorption rather than excretion. Absorption occurs through aspiration, ingestion and burned skin.
- Hepatic damage may progress to failure with hypoprothrombinaemia and hypoglycaemia. Hepatorenal syndrome may occur.
- From intoxication may also result in decreased cardiac output and increased cardiac pooling which subsequently produces hypotension.
- Serum iron should be analysed in symptomatic patients. Serum iron levels (2-4 hrs post-ingestion) greater that 100 ug/dL indicate poisoning with levels, in excess of 350 ug/dL, being potentially serious. Emesis or lavage (for obtunded patients with no gag reflex) are the usual means of decontamination.
- Activated charcoal does not effectively bind iron.
- Catharsis (using sodium sulfate or magnesium sulfate) may only be used if the patient already has diarrhoea.
- Deferoxamine is a specific chelator of ferric (3+) iron and is currently the antidote of choice. It should be administered parenterally. [Ellenhorn and Barceloux: Medical Toxicology]

For acute or short term repeated exposures to xylene:

- Gastro-intestinal absorption is significant with ingestions. For ingestions exceeding 1-2 ml (xylene)/kg, intubation and lavage with cuffed endotracheal tube is recommended. The use of charcoal and cathartics is equivocal.
- ▶ Pulmonary absorption is rapid with about 60-65% retained at rest.
- ▶ Primary threat to life from ingestion and/or inhalation, is respiratory failure.
- Patients should be quickly evaluated for signs of respiratory distress (e.g. cyanosis, tachypnoea, intercostal retraction, obtundation) and given oxygen. Patients with inadequate tidal volumes or poor arterial blood gases (pO2 < 50 mm Hg or pCO2 > 50 mm Hg) should be intubated.
- Arrhythmias complicate some hydrocarbon ingestion and/or inhalation and electrocardiographic evidence of myocardial injury has been reported; intravenous lines and cardiac monitors should be established in obviously symptomatic patients. The lungs excrete inhaled solvents, so that hyperventilation improves clearance.
- A chest x-ray should be taken immediately after stabilisation of breathing and circulation to document aspiration and detect the presence of pneumothorax.
- Epinephrine (adrenalin) is not recommended for treatment of bronchospasm because of potential myocardial sensitisation to catecholamines. Inhaled cardioselective bronchodilators (e.g. Alupent, Salbutamol) are the preferred agents, with aminophylline a second choice.

BIOLOGICAL EXPOSURE INDEX - BEI

These represent the determinants observed in specimens collected from a healthy worker exposed at the Exposure Standard (ES or TLV):

 Determinant
 Index
 Sampling Time
 Comments

 Methylhippu-ric acids in urine
 1.5 gm/gm creatinine
 End of shift

 2 mg/min
 Last 4 hrs of shift

# **SECTION 5 FIREFIGHTING MEASURES**

# Extinguishing media

- Foam
- Dry chemical powder.
- ► BCF (where regulations permit).
- Carbon dioxide.

#### Special hazards arising from the substrate or mixture

Fire Incompatibility Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

# Advice for firefighters

- Alert Fire Brigade and tell them location and nature of hazard.
- Fire Fighting

  May be violently or explosively reactive.
  - ▶ Wear breathing apparatus plus protective gloves.
  - ► Prevent, by any means available, spillage from entering drains or water course.

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Fire/Explosion Hazard

- Liquid and vapour are flammable
- Moderate fire hazard when exposed to heat or flame.
- Vapour forms an explosive mixture with air.
- ▶ Moderate explosion hazard when exposed to heat or flame.

#### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

#### Personal precautions, protective equipment and emergency procedures

**Minor Spills** 

- ► Remove all ignition sources.
- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
- ▶ Control personal contact with the substance, by using protective equipment.

**Major Spills** 

- ► Clear area of personnel and move upwind.
- ▶ Alert Fire Brigade and tell them location and nature of hazard.
- May be violently or explosively reactive.
- ▶ Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

#### **SECTION 7 HANDLING AND STORAGE**

#### Precautions for safe handling

#### Safe handling

- ► Containers, even those that have been emptied, may contain explosive vapours.
- ▶ Do NOT cut, drill, grind, weld or perform similar operations on or near containers.
- DO NOT allow clothing wet with material to stay in contact with skin
- ▶ Electrostatic discharge may be generated during pumping this may result in fire.
- ▶ Ensure electrical continuity by bonding and grounding (earthing) all equipment.

#### Other information

- Store in original containers in approved flammable liquid storage area.
- ► Store away from incompatible materials in a cool, dry, well-ventilated area.
- DO NOT store in pits, depressions, basements or areas where vapours may be trapped.
- ▶ No smoking, naked lights, heat or ignition sources.

#### Conditions for safe storage, including any incompatibilities

#### Suitable container

- ▶ Packing as supplied by manufacturer.
- Plastic containers may only be used if approved for flammable liquid.
- Check that containers are clearly labelled and free from leaks
- For low viscosity materials (i): Drums and jerry cans must be of the non-removable head type.

### Storage incompatibility

- ▶ Vigorous reactions, sometimes amounting to explosions, can result from the contact between aromatic rings and strong oxidising agents. ▶ Aromatics can react exothermically with bases and with diazo compounds.
- ▶ WARNING: Avoid or control reaction with peroxides. All transition metal peroxides should be considered as potentially explosive.

# **SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION**

# **Control parameters**

#### OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	xylene	Xylene (o-, m-, p- isomers)	350 mg/m3 / 80 ppm	655 mg/m3 / 150 ppm	Not Available	Not Available
Australia Exposure Standards	ethylbenzene	Ethyl benzene	434 mg/m3 / 100 ppm	543 mg/m3 / 125 ppm	Not Available	Not Available
Australia Exposure Standards	propylene glycol monomethyl ether acetate, alpha-isomer	1-Methoxy-2-propanol acetate	274 mg/m3 / 50 ppm	548 mg/m3 / 100 ppm	Not Available	Sk

#### **EMERGENCY LIMITS**

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
ferric hydroxide	Iron hydroxide oxide	24 mg/m3	260 mg/m3	1600 mg/m3
xylene	Xylenes	Not Available	Not Available	Not Available
ethylbenzene	Ethyl benzene	Not Available	Not Available	Not Available
propylene glycol monomethyl ether acetate, alpha-isomer	Propylene glycol monomethyl ether acetate, alpha-isomer; (1-Methoxypropyl-2-acetate)	Not Available	Not Available	Not Available
propylene glycol monomethyl ether acetate, beta-isomer	Propylene glycol monomethyl ether acetate, beta-isomer; (2-Methoxypropoyl-1-acetate)	Not Available	Not Available	Not Available

Ingredient	Original IDLH	Revised IDLH
ferric hydroxide	Not Available	Not Available
xylene	1,000 ppm	900 ppm
acrylic polymer	Not Available	Not Available
ethylbenzene	2,000 ppm	800 [LEL] ppm

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propylene glycol monomethyl ether acetate, alpha-isomer	Not Available	Not Available
propylene glycol monomethyl ether acetate, beta-isomer	Not Available	Not Available
Other organic/mineral pigments	Not Available	Not Available

### **Exposure controls**

# Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

# Personal protection











# Eye and face protection

- ► Safety glasses with side shields.
- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task.

#### Skin protection

#### See Hand protection below

- ► Wear chemical protective gloves, e.g. PVC.
- ▶ Wear safety footwear or safety gumboots, e.g. Rubber

### Hands/feet protection

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

#### **Body protection**

#### See Other protection below

#### Other protection

- ► Overalls.
- ► PVC Apron.
- ► PVC protective suit may be required if exposure severe.
- Eyewash unit.

#### Thermal hazards

Not Available

#### Recommended material(s)

# GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

# "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the  $\ computer-$  generated selection:

DUNLOP Concrete Resurfacer Tint – Charcoal

Material	CPI
BUTYL	С
BUTYL/NEOPRENE	С
HYPALON	С
NAT+NEOPR+NITRILE	С
NATURAL+NEOPRENE	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С
PE/EVAL/PE	С
PVA	С
PVC	С
PVDC/PE/PVDC	С
TEFLON	С
VITON	С

<sup>\*</sup> CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

**NOTE**: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

# Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
Protection Factor	Respirator	Respirator	· ·
up to 5 x ES	A-AUS / Class 1	-	A-PAPR-AUS / Class 1
up to 25 x ES	Air-line*	A-2	A-PAPR-2
up to 50 x ES	-	A-3	-
50+ x ES	-	Air-line**	-

 $<sup>^{\</sup>star}$  - Continuous-flow;  $\,^{\star\star}$  - Continuous-flow or positive pressure demand

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

<sup>^ -</sup> Full-face

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#### **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

# Information on basic physical and chemical properties

Appearance	Flammable viscous yellow coloured liquid with characteristic, strong lacquer odour; not miscible with water.		
Physical state	Liquid	Relative density (Water = 1)	1.605-1.615
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	*499 (xylene)
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	136-145	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	23-27	Taste	Not Available
Evaporation rate	0.7 BuAC = 1	Explosive properties	Not Available
Flammability	Flammable.	Oxidising properties	Not Available
Upper Explosive Limit (%)	7.1	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	1.3	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	0.8-1.2	Gas group	Not Available
Solubility in water (g/L)	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	3.7	VOC g/L	Not Available

#### **SECTION 10 STABILITY AND REACTIVITY**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

# **SECTION 11 TOXICOLOGICAL INFORMATION**

ferric hydroxide

Not Available

Information on toxicologic	cal effects		
Inhaled	Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful.  There is some evidence to suggest that the material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.  Headache, fatigue, tiredness, irritability and digestive disturbances (nausea, loss of appetite and bloating) are the most common symptoms of xylene overexposure.		
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.  Swallowing of the liquid may cause aspiration into the lungs with the risk of chemical pneumonitis; serious consequences may result. (ICSC13733)		
Skin Contact	Skin contact with the material may be harmful; systemic effects may result following absorption.  This material can cause inflammation of the skin on contact in some persons.  The material may accentuate any pre-existing dermatitis condition  Repeated exposure may cause skin cracking, flaking or drying following normal handling and use.  Open cuts, abraded or irritated skin should not be exposed to this material  Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects.		
Eye	If applied to the eyes, this material causes severe eye damage.		
Chronic	There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment.  Harmful: danger of serious damage to health by prolonged exposure through inhalation.  This material can cause serious damage if one is exposed to it for long periods. It can be assumed that it contains a substance which can produce severe defects.		
DUNLOP Concrete	TOXICITY	IRRITATION	
Resurfacer Tint – Charcoal	Not Available	Not Available	
ferric hydroxide	TOXICITY	IRRITATION	

Not Available

Respiratory or Skin sensitisation

Mutagenicity

0

0

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	TOXICITY	IRRITATION	
_	Dermal (rabbit) LD50: >1700 mg/kg <sup>[2]</sup>	Eye (human): 2	
xylene	Inhalation (rat) LC50: 5000 ppm/4h <sup>[2]</sup>		ng/24h SEVERE
	Oral (rat) LD50: 4300 mg/kgt <sup>[2]</sup>	Eye (rabbit): 87	
		Skin (raddit):50	0 mg/24h moderate
	TOXICITY	IRRITATION	
	Dermal (rabbit) LD50: ca.15432.6 mg/kg <sup>[1]</sup>	Eye (rabbit): 50	0 mg - SEVERE
ethylbenzene	Inhalation (mouse) LC50: 35.5 mg/L/2H <sup>[2]</sup>	Skin (rabbit): 15	5 mg/24h mild
	Inhalation (rat) LC50: 55 mg/L/2H <sup>[2]</sup>		
	Oral (rat) LD50: 3500 mg/kgd <sup>[2]</sup>		
	TOXICITY	IRRITATION	
propylene glycol	Dermal (rabbit) LD50: >5000 mg/kg*] <sup>[2]</sup>	* [CCINFO]	
monomethyl ether acetate,	Inhalation (rat) LC50: 4345 ppm/6h <sup>[2]</sup>	Nil reported	
alpha-isomer	Oral (rat) LD50: >8532 mg/kgd <sup>[2]</sup>	1	
		I	
	TOXICITY	IRRITATION	
propylene glycol monomethyl ether acetate,	Dermal (rabbit) LD50: >5000 mg/kg** <sup>[2]</sup>	[CCINFO]*	
beta-isomer	Inhalation (rat) LC50: 4345 ppm/6h <sup>[2]</sup>		
	Oral (rat) LD50: 8532 mg/kgd <sup>[2]</sup>		
Legend:	Nalue obtained from Europe ECHA Registered Substance     extracted from RTECS - Register of Toxic Effect of chemica		from manufacturer's SDS. Unless otherwise specified data
	extracted from N1203 - Neglister of Toxic Effect of Chemica	Toubstarioes	
FERRIC HYDROXIDE	No significant acute toxicological data identified in literatu	ıre search.	
	The material may produce severe irritation to the eye caus	sing pronounced inflammation. Repea	ted or prolonged exposure to irritants may produce
XYLENE	conjunctivitis.		ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles,
XYLENE	conjunctivitis.  The material may cause skin irritation after prolonged or re scaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye caus conjunctivitis.	epeated exposure and may produce of the control of	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles,
	conjunctivitis.  The material may cause skin irritation after prolonged or re scaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye caus conjunctivitis.  The material may cause skin irritation after prolonged or re scaling and thickening of the skin.  Ethylbenzene is readily absorbed when inhaled, swallowed	epeated exposure and may produce of the property of a wide variety of propylene glycologing and toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and fetus, blood (haemon toxicities associated with the logoning embryo and the	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles, or contact skin redness, swelling, the production of vesicles, is (musculoskeletal system) recorded.  In-butyl ether (DPnB); dipropylene glycol methyl ether ethers has shown that propylene glycol-based ethers are wer molecular weight homologues of the ethylene series, polytic effects), or thymus, are not seen with the droxyl group produces an alkoxyacetic acid. er) was associated with a teratogenic response in rabbits; wrises only 10% of the commercial material, the remaining
ETHYLBENZENE PROPYLENE GLYCOL	conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye cause conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  Ethylbenzene is readily absorbed when inhaled, swallowed Liver changes, utheral tract, effects on fertility, foetotoxicity  for propylene glycol ethers (PGEs):  Typical propylene glycol ethers include propylene glycol nacetate (DPMA); tripropylene glycol methyl ether (TPM). Testing of a wide variety of propylene glycol ethers. The corsuch as adverse effects on reproductive organs, the develocommercial-grade propylene glycol ethers. In the ethylene A BASF report (in ECETOC) showed that inhalation expect the power of the propylene glycol ethers. In the ethylene A BASF report (in ECETOC) showed that inhalation expect the power of	expeated exposure and may produce of the produce of	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles, on contact skin redness, swelling, the production of vesicles, is (musculoskeletal system) recorded.  In-butyl ether (DPnB); dipropylene glycol methyl ether ethers has shown that propylene glycol-based ethers are wer molecular weight homologues of the ethylene series, oblytic effects), or thymus, are not seen with the droxyl group produces an alkoxyacetic acid. er) was associated with a teratogenic response in rabbits; orises only 10% of the commercial material, the remaining temical. [I.C.I] *Shin-Etsu SDS  asses. This may be due to a non-allergenic condition known levels of highly irritating compound. Key criteria for the dual, with abrupt onset of persistent asthma-like symptoms spirometry, with the presence of moderate to severe inflammation, without eosinophilia, have also been included
PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, ALPHA-ISOMER  PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, BETA-ISOMER	conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye cause conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  Ethylbenzene is readily absorbed when inhaled, swallowed Liver changes, utheral tract, effects on fertility, foetotoxicity for propylene glycol ethers (PGEs):  Typical propylene glycol ethers include propylene glycol in acetate (DPMA); tripropylene glycol methyl ether (TPM). Testing of a wide variety of propylene glycol ethers. The consuch as adverse effects on reproductive organs, the develd commercial-grade propylene glycol ethers. In the ethylene A BASF report (in ECETOC) showed that inhalation expoput exposure to 145 ppm and 36 ppm had no adverse effective airways dysfunction syndrome (RADS) which diagnosis of RADS include the absence of preceding responding hypercactivity on methacholine challenge testing in the criteria for diagnosis of RADS.  No data for material. Data for isomer only as its alpha isomer.	epeated exposure and may produce of a property of a wide variety of propylene glycol monometry of the need for care in handling this characteristics. The beta isomer of PGMEA comps the need for care in handling this characteristics associated with the looping embryo and fetus, blood (haems series, metabolism of the terminal hy soure to 545 ppm PGMEA (beta isom cts. The beta isomer of PGMEA comps the need for care in handling this characteristics are accordingly of the material cears after exposure to the material cears after exposure to the material cears accordingly of the propylene glycol monomethyl ether; propylene glycol monomethyl ether; propylene glycol monomethyl ethers.	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles, in contact skin redness, swelling, the production of vesicles, is (musculoskeletal system) recorded.  In-butyl ether (DPnB); dipropylene glycol-based ethers are wer molecular weight homologues of the ethylene series, blytic effects), or thymus, are not seen with the droxyl group produces an alkoxyacetic acid. er) was associated with a teratogenic response in rabbits; irrises only 10% of the commercial material, the remaining termical. [I.C.I.] *Shin-Etsu SDS  asses. This may be due to a non-allergenic condition known levels of highly irritating compound. Key criteria for the dual, with abrupt onset of persistent asthma-like symptoms spirometry, with the presence of moderate to severe inflammation, without eosinophilia, have also been included er acetate:
PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, ALPHA-ISOMER  PROPYLENE GLYCOL MONOMETHYL ETHER	conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye caus conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  Ethylbenzene is readily absorbed when inhaled, swallowed Liver changes, utheral tract, effects on fertility, foetotoxicity for propylene glycol ethers (PGEs):  Typical propylene glycol ethers (PGEs):  Typical propylene glycol ethers include propylene glycol nacetate (DPMA); tripropylene glycol methyl ether (TPM).  Testing of a wide variety of propylene glycol ethers Testing less toxic than some ethers of the ethylene series. The corsuch as adverse effects on reproductive organs, the devel commercial-grade propylene glycol ethers. In the ethylene A BASF report (in ECETOC) showed that inhalation expound that exposure to 145 ppm and 36 ppm had no adverse effe 90% is alpha isomer. Hazard appears low but emphasize:  Asthma-like symptoms may continue for months or even yeas reactive airways dysfunction syndrome (RADS) which diagnosis of RADS include the absence of preceding responding in the criteria for diagnosis of RADS.	epeated exposure and may produce of the produce of	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles, in contact skin redness, swelling, the production of vesicles, is (musculoskeletal system) recorded.  In-butyl ether (DPnB); dipropylene glycol-based ethers are wer molecular weight homologues of the ethylene series, plytic effects), or thymus, are not seen with the droxyl group produces an alkoxyacetic acid. er) was associated with a teratogenic response in rabbits; virses only 10% of the commercial material, the remaining remical. [I.C.I] *Shin-Etsu SDS  ases. This may be due to a non-allergenic condition known levels of highly irritating compound. Key criteria for the dual, with abrupt onset of persistent asthma-like symptoms spirometry, with the presence of moderate to severe inflammation, without eosinophilia, have also been included er acetate:
PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, ALPHA-ISOMER  PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, BETA-ISOMER  Acute Toxicity	conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Reproductive effector in rats  The material may produce severe irritation to the eye caus conjunctivitis.  The material may cause skin irritation after prolonged or rescaling and thickening of the skin.  Ethylbenzene is readily absorbed when inhaled, swallowed Liver changes, utheral tract, effects on fertility, foetotoxicity for propylene glycol ethers (PGEs):  Typical propylene glycol ethers include propylene glycol in acetate (DPMA); tripropylene glycol methyl ether (TPM). Testing of a wide variety of propylene glycol ethers Testing less toxic than some ethers of the ethylene series. The corsuch as adverse effects on reproductive organs, the develor commercial-grade propylene glycol ethers. In the ethylene A BASF report (in ECETOC) showed that inhalation exposure to 145 ppm and 36 ppm had no adverse effects on the sposure to texposure to 145 ppm and 36 ppm had no adverse effects on the sposure to the sposure to take the sposure to take the sposure to the sp	epeated exposure and may produce of a property of a wide variety of propylene glycol monometry of the need for care in handling this characteristics. The beta isomer of PGMEA comps the need for care in handling this characteristics associated with the looping embryo and fetus, blood (haems series, metabolism of the terminal hy soure to 545 ppm PGMEA (beta isom cts. The beta isomer of PGMEA comps the need for care in handling this characteristics are accordingly of the material cears after exposure to the material cears after exposure to the material cears accordingly of the propylene glycol monomethyl ether; propylene glycol monomethyl ether; propylene glycol monomethyl ethers.	ted or prolonged exposure to irritants may produce in contact skin redness, swelling, the production of vesicles, in contact skin redness, swelling, the production of vesicles, is (musculoskeletal system) recorded.  In-butyl ether (DPnB); dipropylene glycol methyl ether ethers has shown that propylene glycol-based ethers are wer molecular weight homologues of the ethylene series, blytic effects), or thymus, are not seen with the droxyl group produces an alkoxyacetic acid. er) was associated with a teratogenic response in rabbits; irrises only 10% of the commercial material, the remaining termical. [I.C.I.] *Shin-Etsu SDS  asses. This may be due to a non-allergenic condition known levels of highly irritating compound. Key criteria for the dual, with abrupt onset of persistent asthma-like symptoms spirometry, with the presence of moderate to severe inflammation, without eosinophilia, have also been included er acetate:

Aspiration Hazard

STOT - Repeated Exposure

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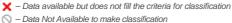
# **DUNLOP Concrete Resurfacer Tint – Charcoal**

Issue Date: 10/08/2015 Print Date: 12/08/2015

**Legena**:

**SECTION 12 ECOLOGICAL INFORMATION** 

🔻 — Data requireu to mane ciassilication avaliable



Toxicity

Toxic to aquatic organisms.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
xylene	HIGH (Half-life = 360 days)	LOW (Half-life = 1.83 days)
ethylbenzene	HIGH (Half-life = 228 days)	LOW (Half-life = 3.57 days)
propylene glycol monomethyl ether acetate, alpha-isomer	LOW	LOW
propylene glycol monomethyl ether acetate, beta-isomer	LOW	LOW

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation
xylene	MEDIUM (BCF = 740)
ethylbenzene	LOW (BCF = 79.43)
propylene glycol monomethyl ether acetate, alpha-isomer	LOW (LogKOW = 0.56)
propylene glycol monomethyl ether acetate, beta-isomer	LOW (LogKOW = 0.5163)

# Mobility in soil

Ingredient	Mobility
ethylbenzene	LOW (KOC = 517.8)
propylene glycol monomethyl ether acetate, alpha-isomer	HIGH (KOC = 1.838)
propylene glycol monomethyl ether acetate, beta-isomer	HIGH (KOC = 1.838)

### **SECTION 13 DISPOSAL CONSIDERATIONS**

# Waste treatment methods

Product / Packaging disposal

- ► Containers may still present a chemical hazard/ danger when empty.
- ▶ Return to supplier for reuse/ recycling if possible.

- F If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- ▶ Where possible retain label warnings and MSDS and observe all notices pertaining to the product.

# **SECTION 14 TRANSPORT INFORMATION**

# **Labels Required**



Marine Pollutant



HAZCHEM

# Land transport (ADG)

UN number	1263
Packing group	III
UN proper shipping name	PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)
Environmental hazard	No relevant data

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Transport hazard class(es)	Class 3 Subrisk Not Applicable
Special precautions for user	Special provisions 163 223 *  Limited quantity 5 L

# Air transport (ICAO-IATA / DGR)

1181	1000	
UN number	1263	
Packing group	III	
UN proper shipping name	Paint (including paint, lacquer, enamel, stain, shellac, varnish, preducing compounds)	olish, liquid filler and liquid lacquer base); Paint related material (including paint thinning or
Environmental hazard	No relevant data	
Transport hazard class(es)	ICAO/IATA Class 3	
	ICAO / IATA Subrisk Not Applicable	
	ERG Code 3L	
Special precautions for user	Special provisions	A3 A72 A192
	Cargo Only Packing Instructions	366
	Cargo Only Maximum Qty / Pack	220 L
	Passenger and Cargo Packing Instructions	355
	Passenger and Cargo Maximum Qty / Pack	60 L
	Passenger and Cargo Limited Quantity Packing Instructions	Y344
	Passenger and Cargo Limited Maximum Qty / Pack	10 L

# Sea transport (IMDG-Code / GGVSee)

UN number	1263	
Packing group		
UN proper shipping name	PAINT (including paint, lacquer, enamel, stain, shellac, varnish, polish, liquid filler and liquid lacquer base) or PAINT RELATED MATERIAL (including paint thinning or reducing compound)	
Environmental hazard	Not Applicable	
Transport hazard class(es)	IMDG Class 3 IMDG Subrisk Not Applicable	
Special precautions for user	EMS Number F-E , S-E Special provisions 163 223 955 Limited Quantities 5 L	

# Transport in bulk according to Annex II of MARPOL 73 / 78 and the IBC code

Source	Ingredient	Pollution Category
IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk	xylene	Υ
IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk	ethylbenzene	Y
IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk	propylene glycol monomethyl ether acetate, alpha-isomer	Z

# **SECTION 15 REGULATORY INFORMATION**

Safety, health and environmental regulations / legislation specific for the substance or mixture

FERRIC HYDROXIDE(20344-49-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Inventory of Chemical Substances (AICS)

Australia Exposure Standards

XYLENE(1330-20-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances Information System - Consolidated Lists

Australia Inventory of Chemical Substances (AICS) International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

Monographs

ETHYLBENZENE(100-41-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

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#### **DUNLOP Concrete Resurfacer Tint - Charcoal**

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Australia Exposure Standards Australia Inventory of Chemical Substances (AICS) Australia Hazardous Substances Information System - Consolidated Lists International Agency for Research on Cancer (IARC) - Agents Classified by the IARC

Monographs

#### PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, ALPHA-ISOMER(108-65-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Exposure Standards Australia Inventory of Chemical Substances (AICS) Australia Hazardous Substances Information System - Consolidated Lists

#### PROPYLENE GLYCOL MONOMETHYL ETHER ACETATE, BETA-ISOMER(70657-70-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances Information System - Consolidated Lists Australia Inventory of Chemical Substances (AICS)

National Inventory	Status
Australia - AICS	Υ
Canada - DSL	Υ
Canada - NDSL	N (propylene glycol monomethyl ether acetate, alpha-isomer; ferric hydroxide; xylene; ethylbenzene; propylene glycol monomethyl ether acetate, beta-isomer)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	N (ferric hydroxide; propylene glycol monomethyl ether acetate, beta-isomer)
Korea - KECI	Υ
New Zealand - NZIoC	Υ
Philippines - PICCS	Υ
USA - TSCA	N (propylene glycol monomethyl ether acetate, beta-isomer)
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

### **SECTION 16 OTHER INFORMATION**

#### Other information

#### Ingredients with multiple cas numbers

Name	CAS No
ferric hydroxide	135507-54-9, 20344-49-4, 681125-92-8, 741267-27-6
propylene glycol monomethyl ether acetate, alpha-isomer	108-65-6, 142300-82-1, 84540-57-8

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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